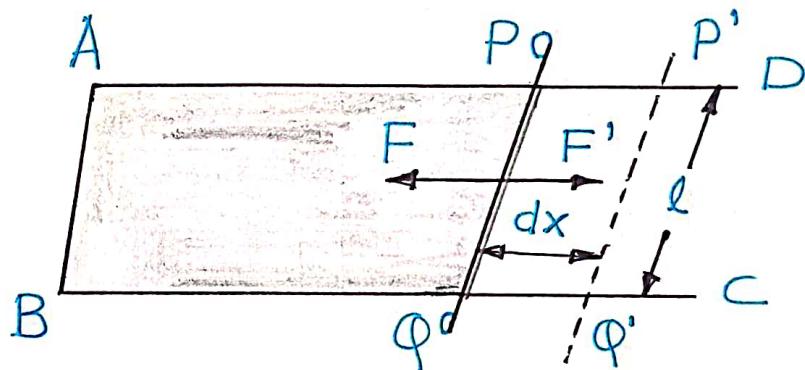


Relation between surface tension & surface energy per unit area.

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Show that the surface tension of a liquid is numerically equal to the surface energy per unit area.

Diagram :-



- consider a rectangular frame ABCD fitted with a movable arm PQ .
- The metallic frame is held in horizontal position is dipped into soap solution & taken out so that a soap film $APQB$ is formed.

- Due to surface tension of soap solution, a force 'F' will act on each arm of the frame.
- Under the action of this force, the movable arm PQ moves towards AB.

Now, work done will be,

$$W = F \cdot x \quad \dots \dots (1)$$

But,

$$\text{Surface tension} = \frac{\text{Force}}{\text{Length}}$$

$$T = \frac{F}{l}$$

$$T = \frac{F}{2l}$$

$$F = 2lT$$

Put $F = 2lT$ in eqn (1),

$$W = 2lT \cdot x$$

But this work is stored as surface energy,

$$\text{Surface energy} = \text{workdone}$$

$$S.E. = 2lT \cdot x$$

$$[E = Tq] \cdots [a = 210c]$$

If $a = 1$

$$[T = E]$$